

(16) Today

^{13}C NMR and Practice

Spring Break Begins at 4:30

Next Class (17)

Chap 12: Infrared Spectroscopy

Chap 20

(18) Second Class from Today

Chap 20

Third Class from Today (19)

Chap 20

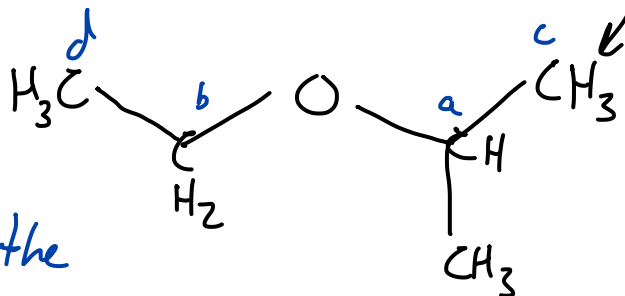
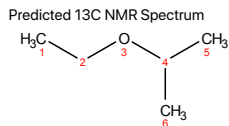
Please rework Test 1 and hand in on March 19.

^1H little magnets 99+ %

^{13}C also little magnets weaker magnets than ^1H
also ~1% of C is ^{13}C

don't see ^{13}C resonance in ^1H NMR because the
resonance frequencies are different

$^{13}\text{C} \{^1\text{H}\}$ NMR

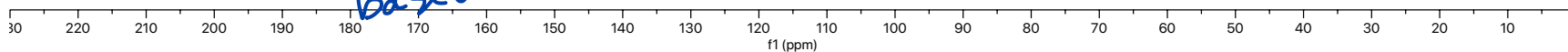


this C has 3 H's bound to it... so this should be a quartet, right? Because we make it not

these are singlets

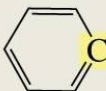
blast the radio of the 'H's so they keep flipping; thus, we decouple the 'H atoms from the ^{13}C atoms ... and singlets result

different chemical environments
 # clues about the chemical environment based on chemical shift



C atoms are in 4 different chemical environments

$^{13}\text{C} \{^1\text{H}\}$ NMRTable 14.4 Approximate Values of Chemical Shifts for ^{13}C NMR

Type of carbon	Approximate chemical shift (ppm)	Type of carbon	Approximate chemical shift (ppm)
$(\text{CH}_3)_4\text{Si}$	0	$\text{C}-\text{I}$	0–40
$\text{R}-\text{CH}_3$	8–35	$\text{C}-\text{Br}$	25–65
$\text{R}-\text{CH}_2-\text{R}$	15–50	$\text{C}-\text{Cl}$	35–80
$\begin{array}{c} \text{R} \\ \\ \text{R}-\text{CH}-\text{R} \end{array}$	20–60	$\text{C}-\text{N}$	40–60
$\begin{array}{c} \text{R} \\ \\ \text{R}-\text{C}-\text{R} \\ \\ \text{R} \end{array}$	30–40	$\text{C}-\text{O}$	50–80
$\equiv\text{C}$	65–85	$\begin{array}{c} \text{R} \\ \diagdown \\ \text{C}=\text{O} \\ \diagup \\ -\text{N} \end{array}$	165–175
$=\text{C}$	100–150	$\begin{array}{c} \text{R} \\ \diagdown \\ \text{C}=\text{O} \\ \diagup \\ \text{RO} \end{array}$	165–175
	110–170	$\begin{array}{c} \text{R} \\ \diagdown \\ \text{C}=\text{O} \\ \diagup \\ \text{HO} \end{array}$	175–185
		$\begin{array}{c} \text{R} \\ \diagdown \\ \text{C}=\text{O} \\ \diagup \\ \text{H} \end{array}$	190–200
		$\begin{array}{c} \text{R} \\ \diagdown \\ \text{C}=\text{O} \\ \diagup \\ \text{R} \end{array}$	205–220

Practice

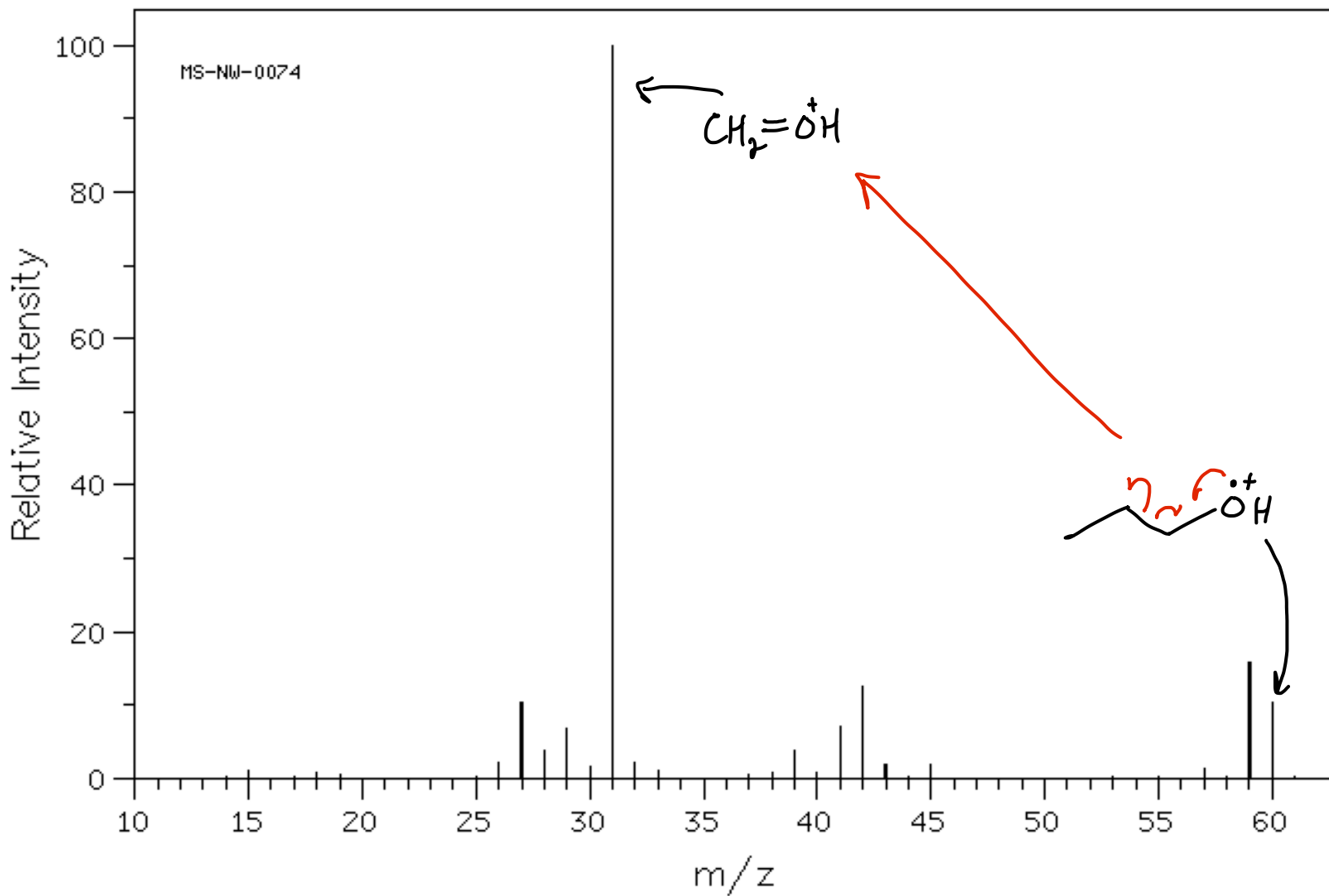
MS

mass of molecule to help determine formula

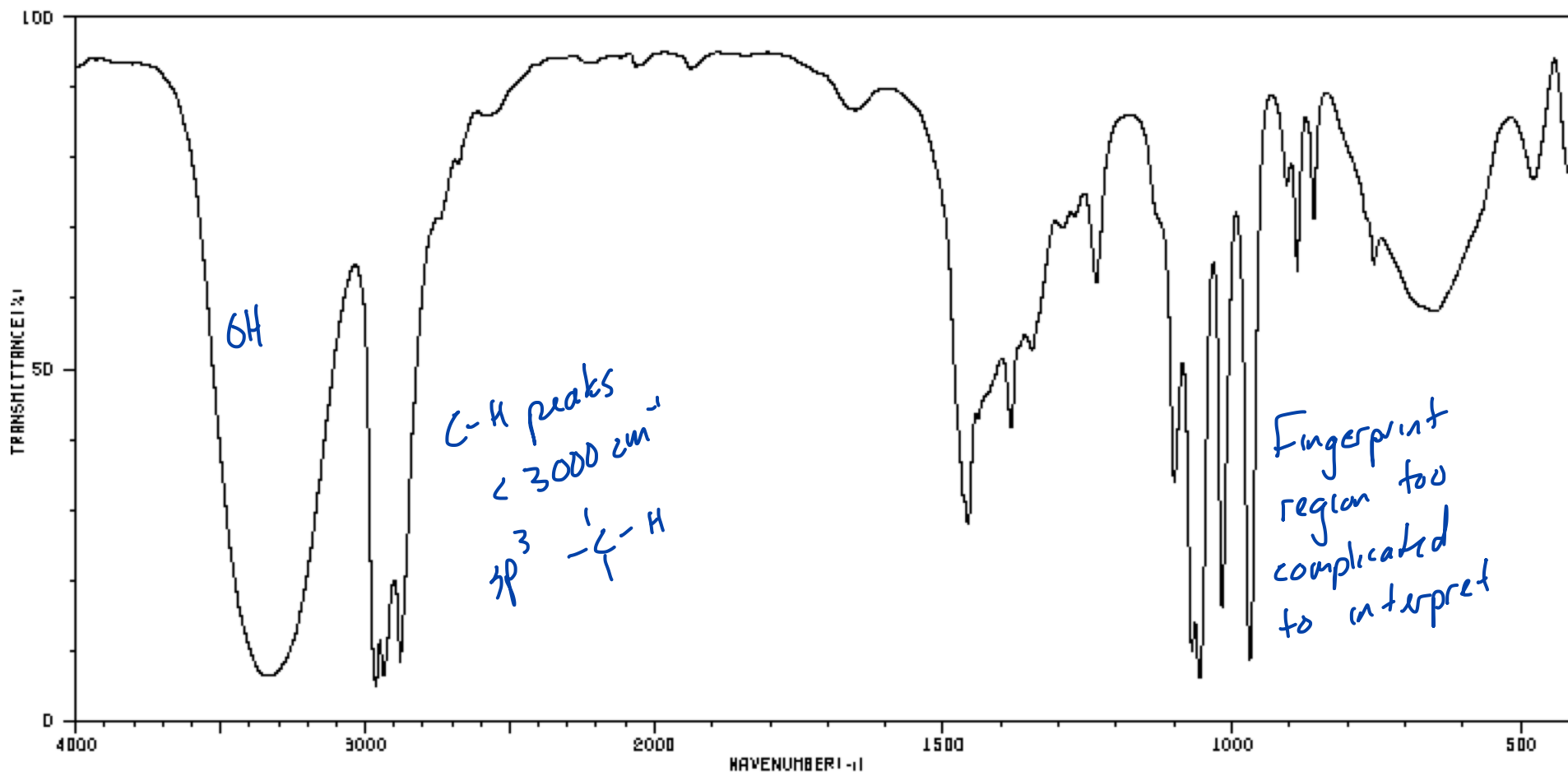
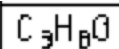
molecular ion = 60

relative
m/z intensity

15	1
26	2
27	10
28	3
29	6
30	1
31	100
32	2
33	1
39	3
41	7
42	12
43	2
45	1
57	1
59	15
60	10



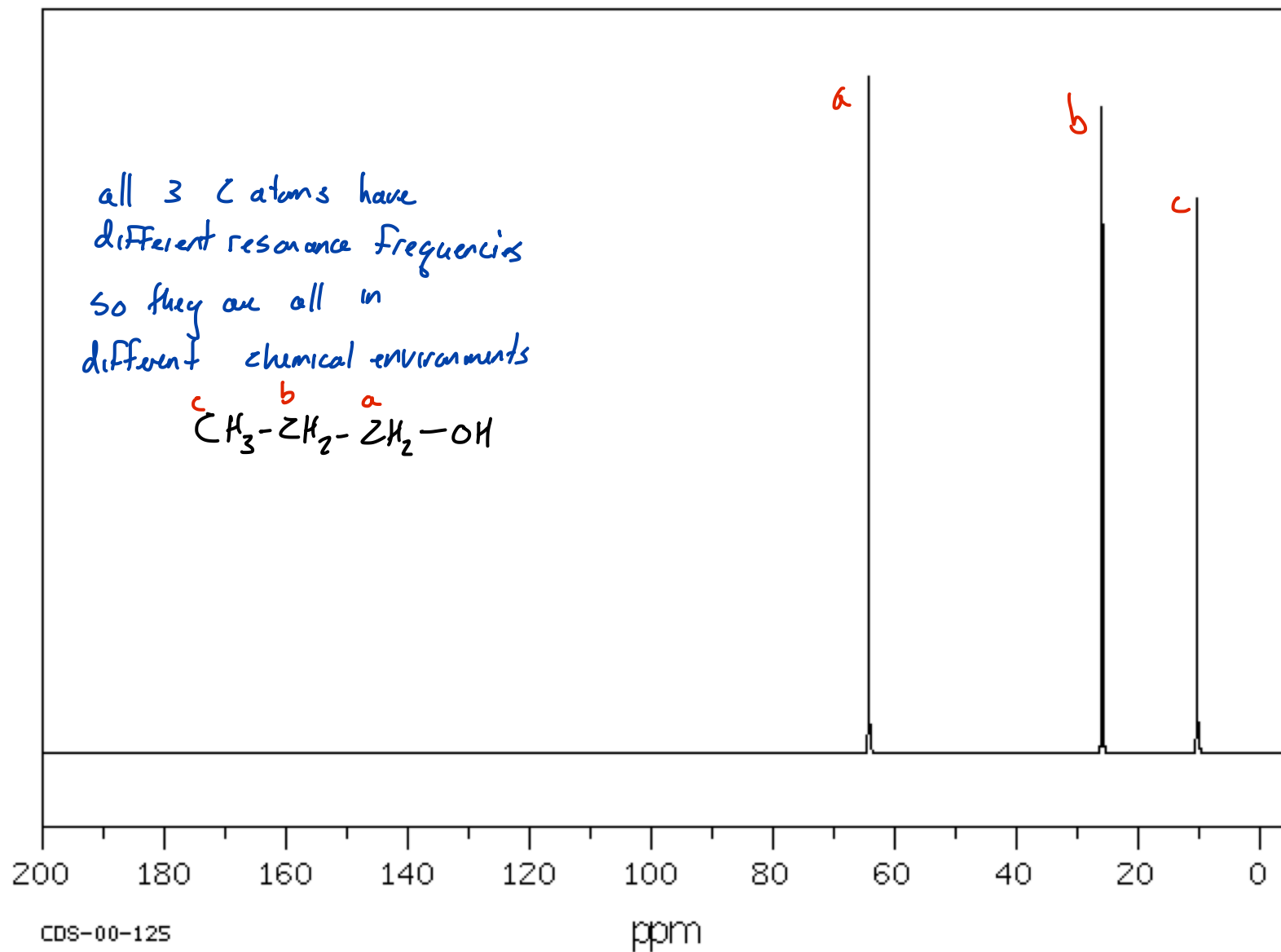
IR



3333	6	1383	39	1017	16
2963	4	1346	50	969	8
2938	6	1293	68	905	72
2878	8	1236	60	888	62
1656	84	1100	32	858	68
1651	84	1069	9	755	82
1466	26	1066	6	479	74

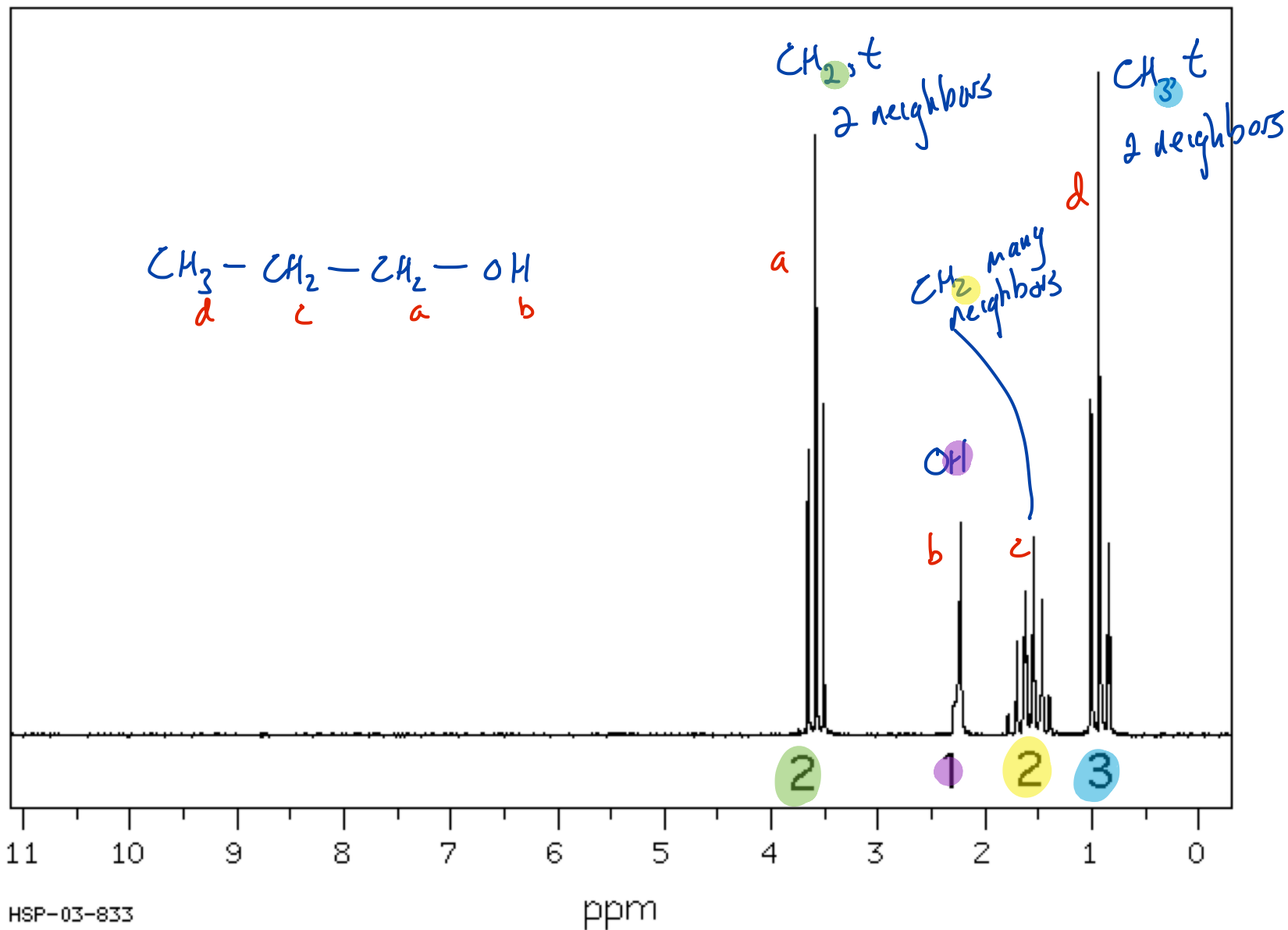


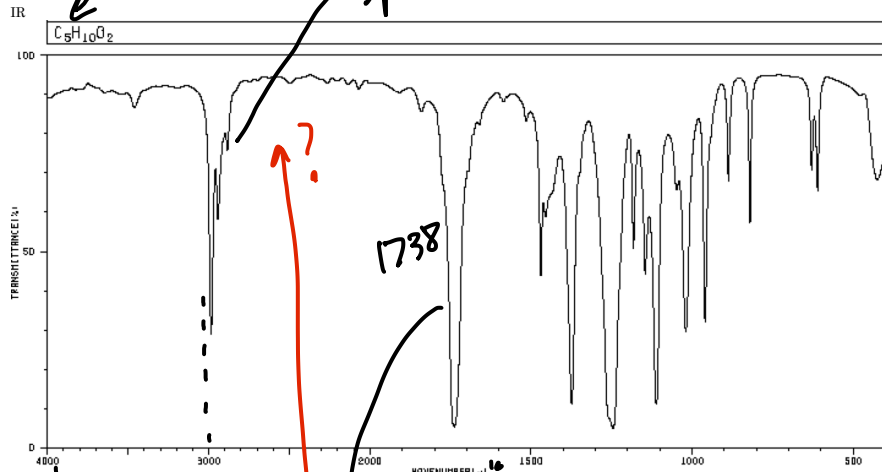
$^{13}\text{C}\{^1\text{H}\}$ NMR



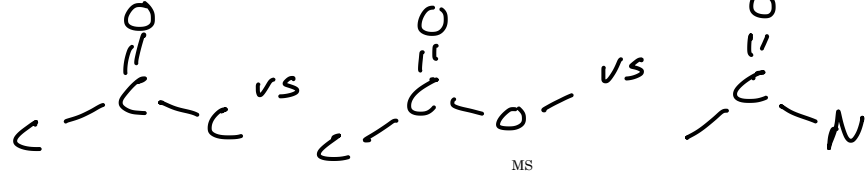
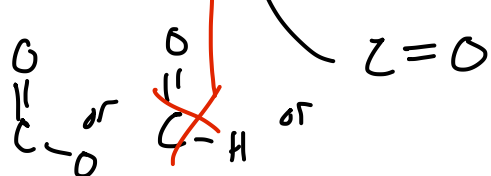
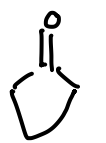
¹H NMR

ppm	Int.
3.658	431
3.585	906
3.511	501
2.225	322
1.717	49
1.699	143
1.629	216
1.544	300
1.467	204
1.46	103
1.398	60
1.009	508
0.932	1000
0.846	289



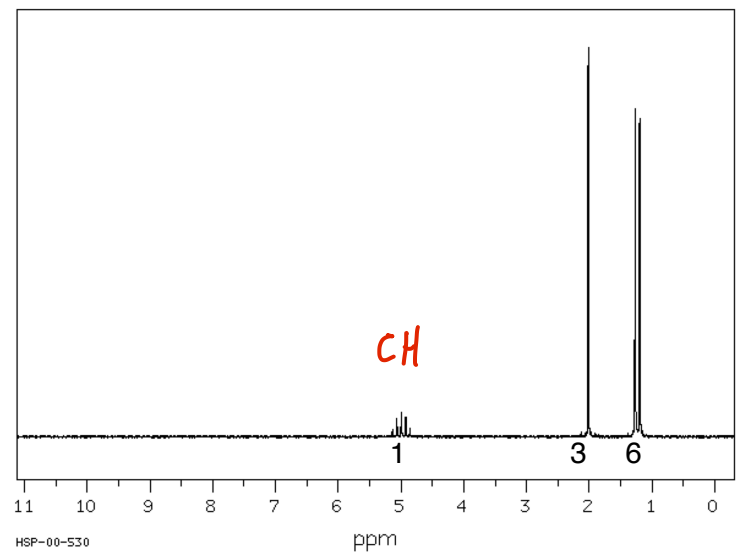


Wavenumber (cm⁻¹)	Intensity	Wavenumber (cm⁻¹)	Intensity
3160	84	1684	84
2984	27	1614	79
2944	55	1470	42
2884	72	1456	57
1738	5	1248	4
1680	86	1183	48
1146	42	821	56
629	58		
511	62		
426	56		



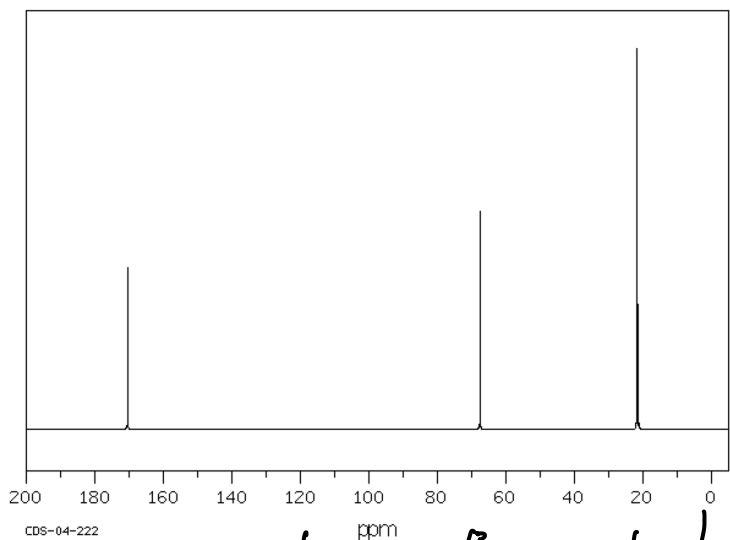
1H NMR

ppm	Int.
5.134	17
5.063	47
4.993	64
4.923	50
4.853	19
2.017	1000
1.268	842
1.198	818



^{13}C NMR

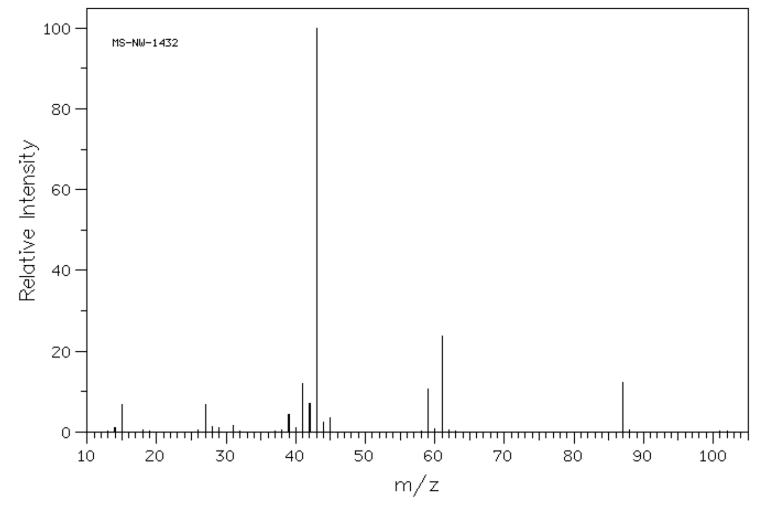
ppm	Int.
170.42	425
67.52	570
21.79	1000
21.34	325

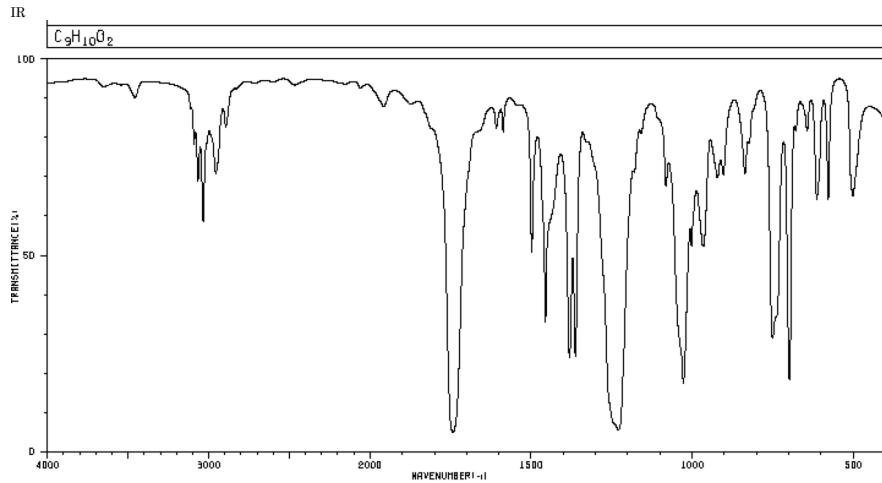


only 3 ^{13}C peaks!

MS

m/z	relative intensity
14	1
15	6
27	6
28	1
29	1
31	1
39	4
40	1
41	12
42	7
43	100
44	2
45	3
59	10
61	23
87	12

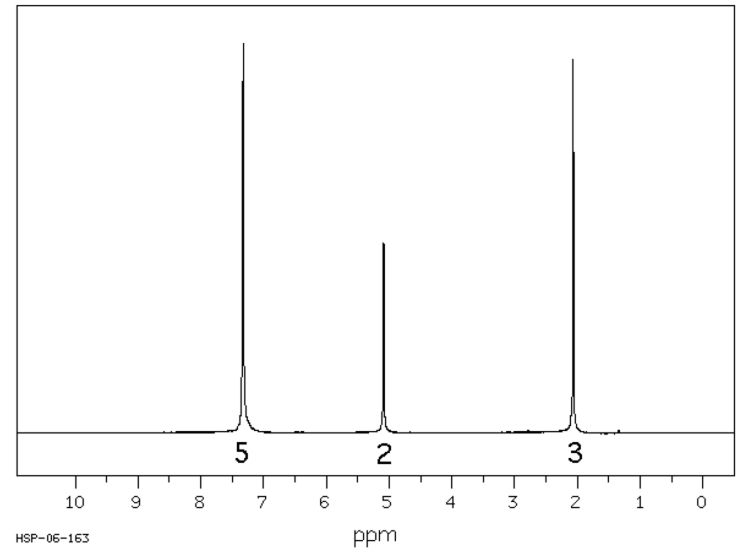




3091	74	1608	79	1081	64	824	74	576	52
3067	66	1587	79	1027	16	751	27	502	62
3035	57	1498	48	1003	50	739	33	487	72
2966	68	1468	31	966	60	698	17		
2895	79	1381	23	922	66	679	79		
1956	84	1363	23	903	88	644	79		
1743	4	1229	5	837	88	614	62		

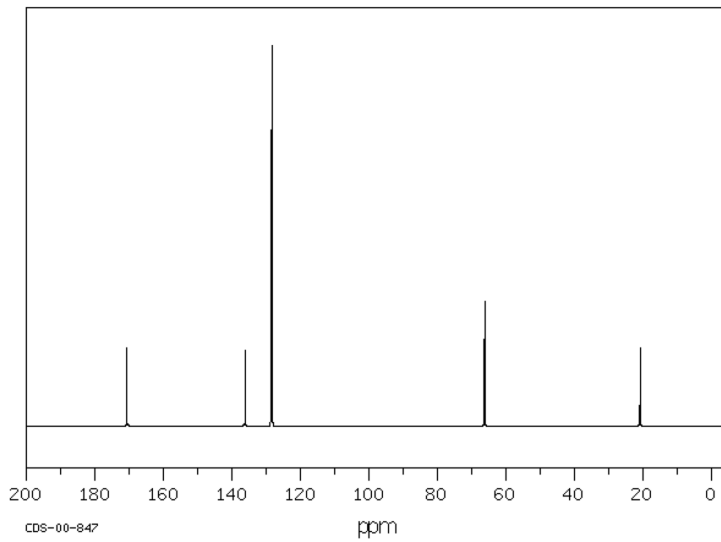
¹H NMR

ppm	Int.
7.327	1000
5.086	488
2.065	960



¹³C{¹H} NMR

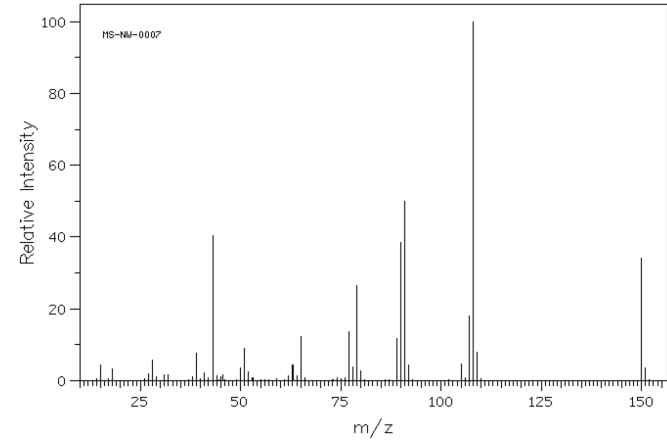
ppm	Int.
170.7	204
136.14	199
128.56	776
128.24	1000
66.24	328
20.82	204

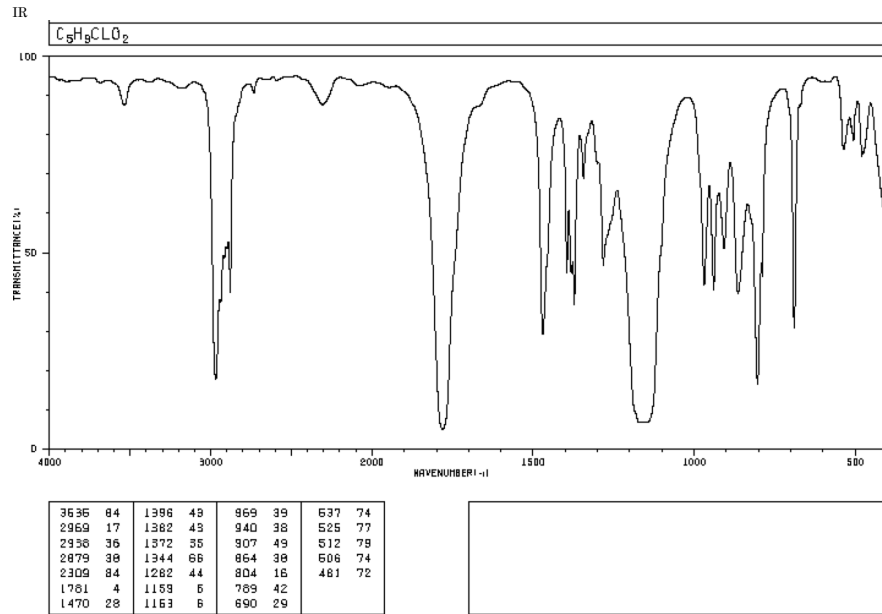


molecular ion = 150

m/z	relative intensity	m/z	relative intensity
15	4	90	38
18	3	91	50
27	1	92	4
28	5	105	4
29	1	107	18
31	1	108	100
32	1	109	7
38	1	150	34
39	7	151	3
41	2		
43	40		
44	1		
45	1		
45.5	1		
50	3		
51	9		
52	2		
62	1		
63	4		
64	1		
65	12		
77	13		
78	3		
79	26		
80	2		
89	11		

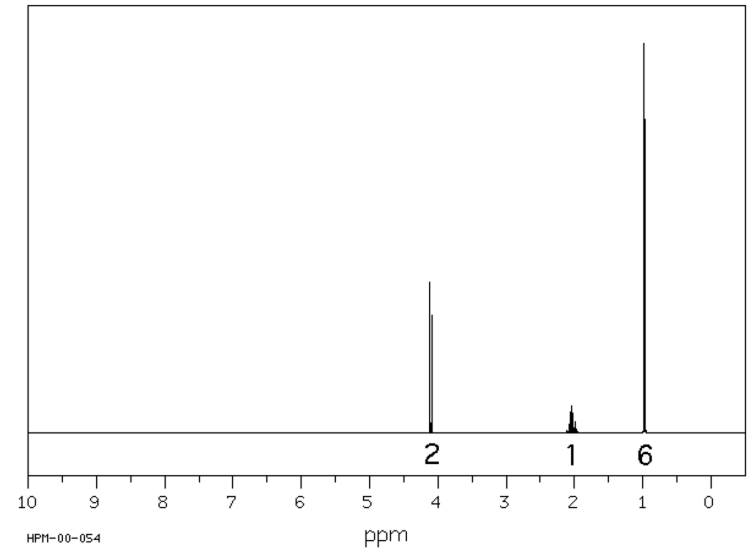
MS



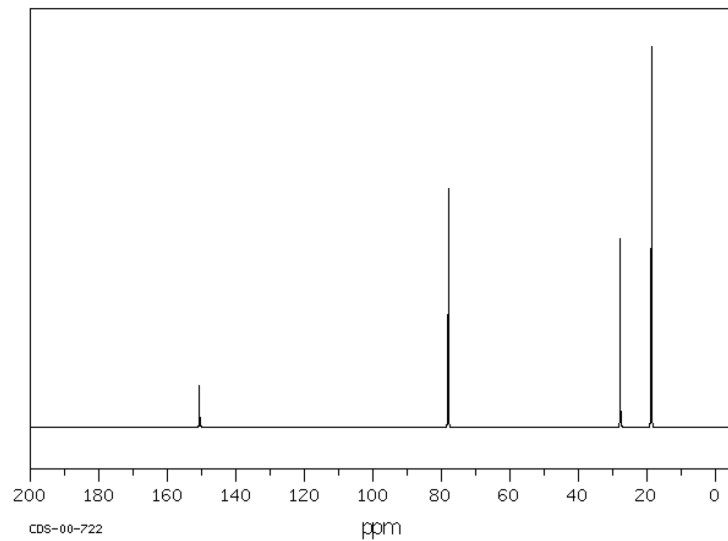


¹H NMR

ppm	Int.
4.118	335
4.096	343
2.113	10
2.091	32
2.069	62
2.046	79
2.024	65
2.002	35
1.98	11
0.992	1000
0.97	957



¹³C{¹H} NMR



molecular ion = 136

m/z	relative intensity
15	1
26	1
27	21
28	4
29	21
31	4
36	1
38	1
39	15
40	3
41	71
42	16
43	100
44	6
53	1
55	8
56	76
57	33
58	1
59	1
63	37
65	12
93	1
94	8
95	4
96	2
97	1

MS

