

Today

Next Class

Sections 4.3 - 4.8  
Chirality

Sections 4.9-4.14  
Optical activity and compounds with more than one  
center of chirality

## Determining Configuration (R vs S)

*same rules as Z/E rules but do 1<sup>st</sup>, 2<sup>nd</sup>, 3<sup>rd</sup>, and 4<sup>th</sup> place*

Assign priorities to groups connected to chirality center

Point lowest priority group away

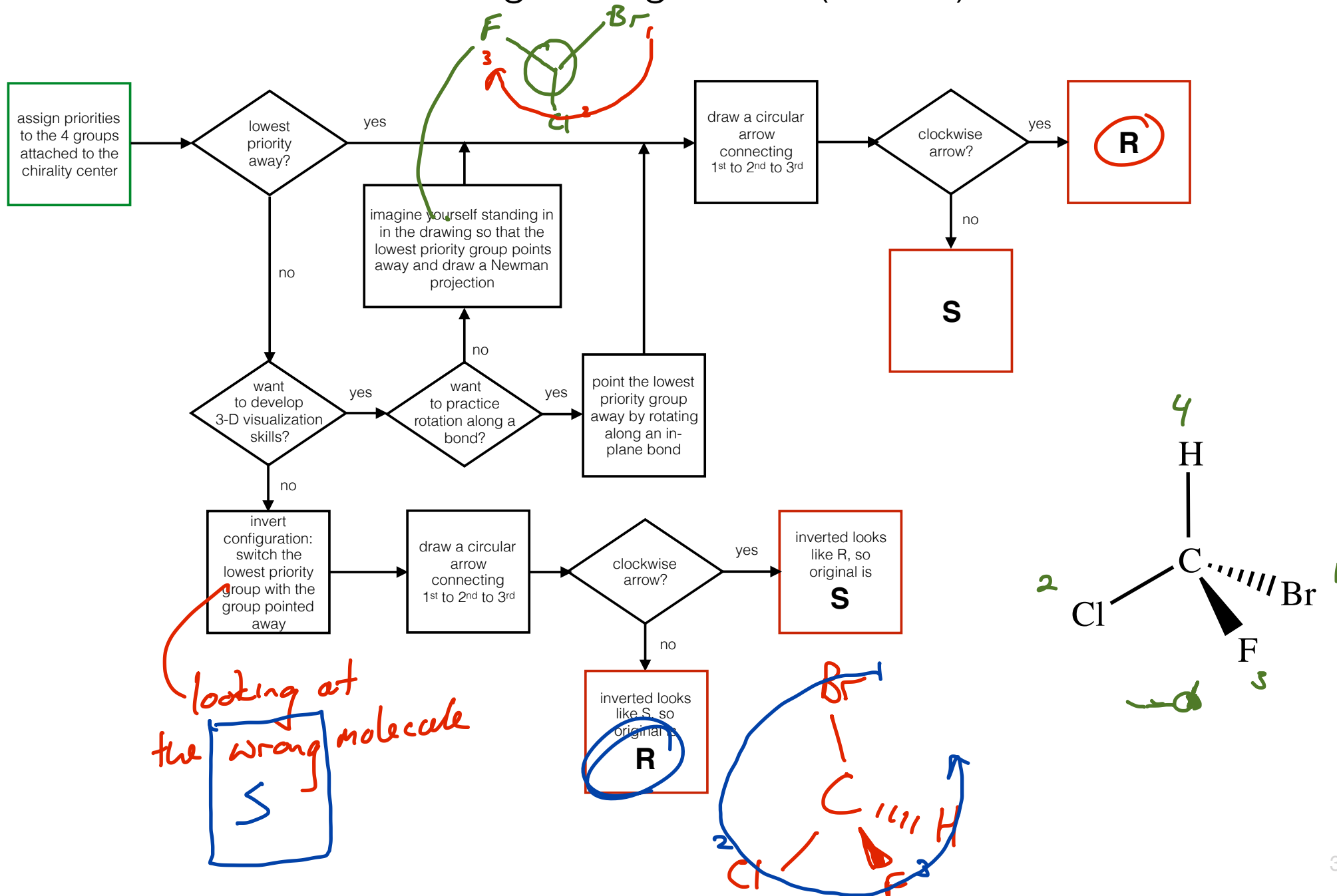
Draw a circle from 1<sup>st</sup> to 2<sup>nd</sup> to 3<sup>rd</sup> priority groups

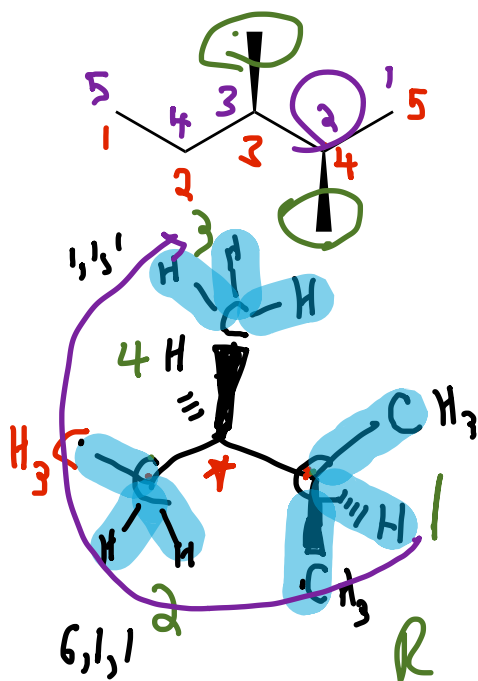
**Clockwise** circle is **R** configuration

**Counter Clockwise** circle is **S** configuration

*remember, it is based on atomic #*

# Determining Configuration (R vs S)

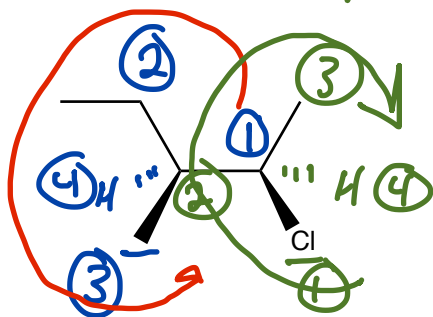
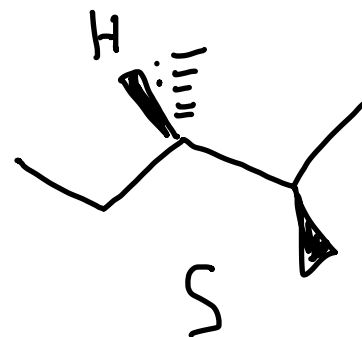




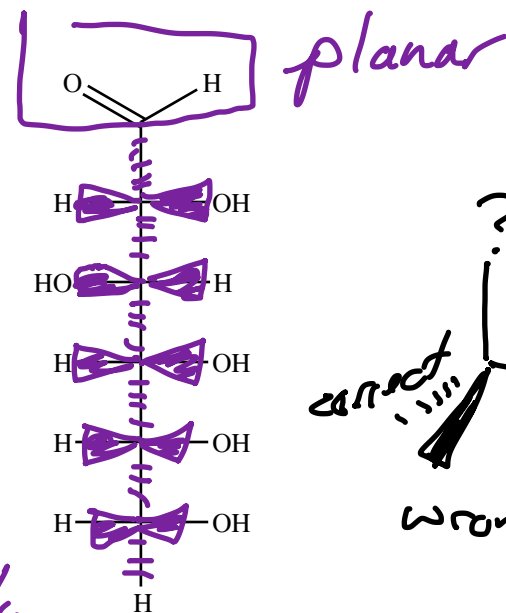
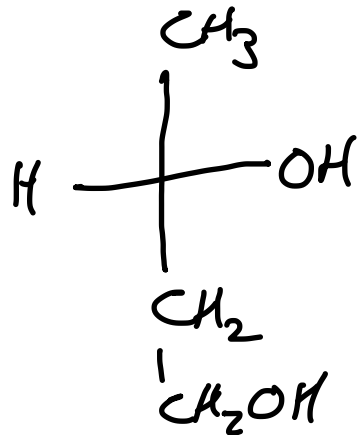
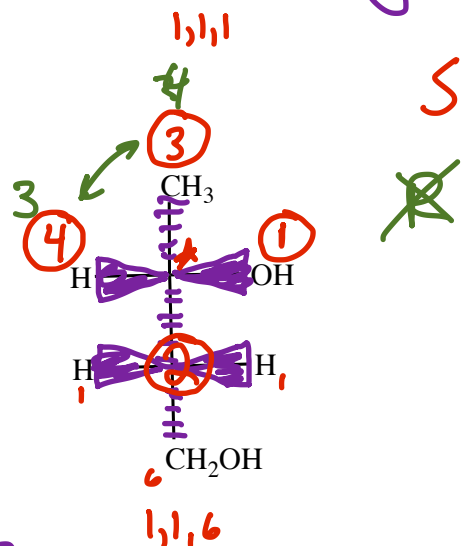
(R) 2,3-dimethylpentane

R  
6,1,1

switching  
two groups  
inverts  
configuration



(2R,3S)-2-chloro-3-methylpentane



Chain of C atoms is drawn vertically

horizontal bonds are coming out of the plane towards the viewer

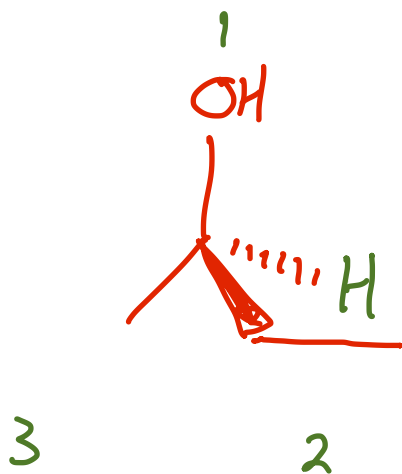
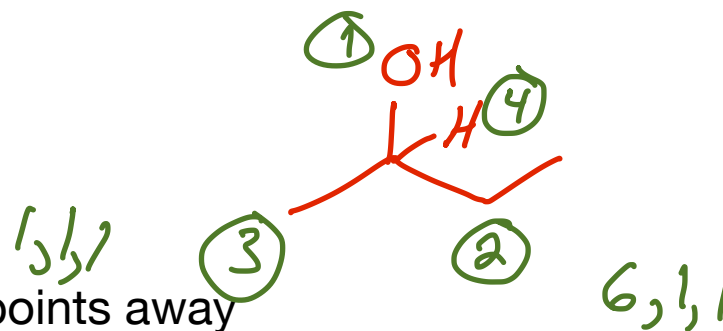
vertical bonds are going out of the plane away from the viewer

If lowest priority group is vertical 1 → 2 → 3 correct answer

If lowest priority group is horizontal 1 → 2 → 3 wrong so report other choice

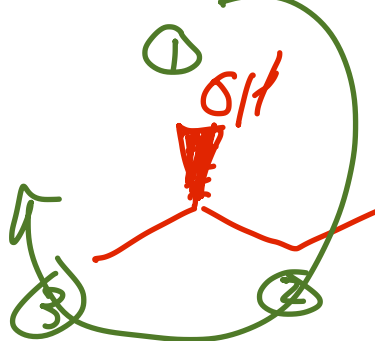
# R-2-butanol

1. Draw a tetrahedral C atom
2. Assign priorities to the groups
3. Place the lowest priority group so that it points away
4. Draw in priority groups 1 through 3 in the correct (clockwise or counterclockwise) orientation.



## Drawing Chiral Molecules (How you really do it)

Section 4.7

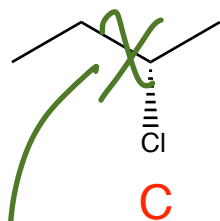
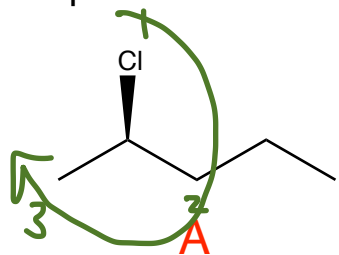


1. Draw the molecule
2. Assign priorities and check if the correct configuration is drawn
3. a. If correct, celebrate, you're done
3. b. If incorrect version is drawn, redraw molecule shaping the positions of 2 (and only two) substituents.

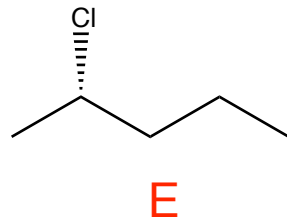
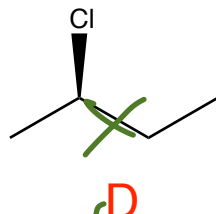
*R*-2-chloropentane

(2*S*,3*S*)-2-bromo-3-chloropentane

R-2-chloropentane



C




E

4 atoms  
butane



1. Draw the molecule
2. Assign priorities and check if the correct configuration is drawn
3. a. If correct, celebrate, you're done
3. b. If incorrect version is drawn, redraw molecule shaping the positions of 2 (and only two) substituents.

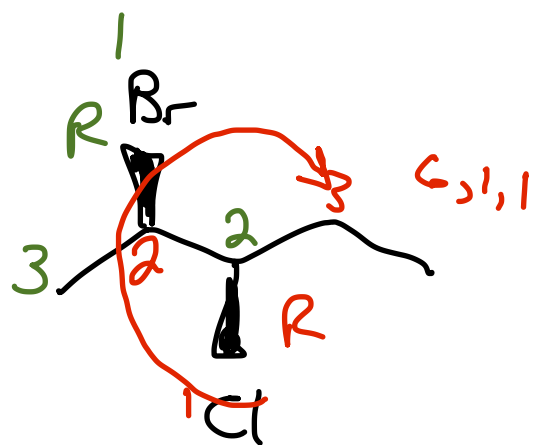
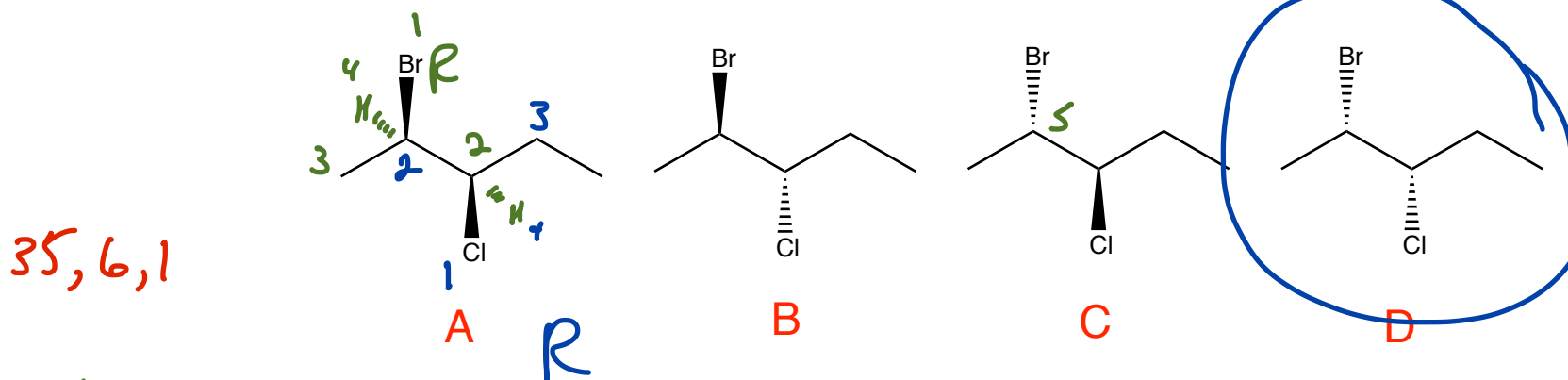
(2S,3S)-2-bromo-3-chloropentane

E + Z are for   
stereoisomers (diastereomers)  
of double bonds

R + S are for configurations  
of chiral atoms

R-2-chloropentane

(2S,3S)-2-bromo-3-chloropentane



$2^n$  possible stereoisomers  
 where  $n$  = number of  
 stereocenters

17, 6, 1